

Lecture 12. Global biogeochemical cycles of carbon.

Objectives:

1. Global biogeochemical cycle of carbon.
2. Important carbon-containing compounds.

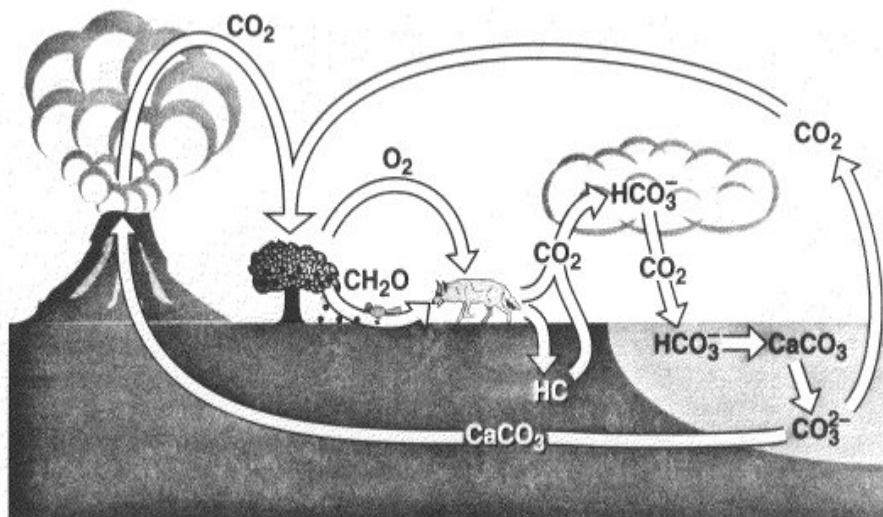
Readings: Turco: p. 307-316; Brimblecombe: p. 21-23, 37 (20-41)

1. Global biogeochemical cycle of carbon

Principal carbon compounds in the atmosphere:

CO_2 , CO , CH_4 , carbonates (e.g., CaCO_3), organic compounds (e.g., CH_2O)

Figure 12.1 The simplified illustration of transfer of carbon through the biosphere and geosphere (Turco 1997).



NOTE: find major differences in Figure 10.14 from Turco (1997) and Figure 2.1 from Brimblecombe (1996)

The carbon cycle on Earth can be divided on the primary (**fast**) processes and the secondary (**slow**) processes.

Major primary (**fast**) carbon transfer processes:

- * Atmosphere/land-biomass exchange of carbon dioxide by **photosynthesis** and **respiration**, 80Gt-C/yr.
- * Atmosphere/ocean exchange of CO₂ by **dissolution** and **evaporation**, 50Gt-C/yr.
- * Surface-ocean/deep-ocean exchange of carbonate by **sinking** and **upwelling** of ocean water, 25Gt-C/yr.
- * Surface-ocean/water-biomass exchange of CO₂ by **photosynthesis** and **respiration**, 5Gt-C/yr.

Major secondary (**slow**) carbon transfer processes:

- * Sediment/atmosphere transfer of carbonate as CO₂ by silicate mineral reconstruction and **volcanism**, 0.05Gt-C/yr.
- * Sediment/ocean transfer of carbonate by **weathering** and **runoff** from land, 0.2Gt-C/yr.
- * Ocean/sediment transfer of carbonate by **sedimentation** and burial of tests, 0.25Gt-C/yr.
- * Sediment/atmosphere transfer of fossilized organic carbon by oxidation during **weathering** of uplifted sediments, 0.05Gt-C/yr.
- * Ocean/sediment transfer of organic carbon by **sedimentation** and burial of detritus, 0.05Gt-C/yr.

2. Important carbon-containing compounds:

CO₂, CO, CH₄, carbonates (e.g., CaCO₃), organic compounds (e.g., CH₂O)

- Carbon dioxide is the single most important trace constituent from the standpoint of global climate change causing global warming (will be discussed in Lectures 33-34). CO₂ levels have increased from about 280 ppm in 1800 to 356 ppm in 1993.
- Carbon monoxide has a chemical lifetime of 30 to 90 days on the global scale of the troposphere. Tropospheric CO mixing ratios range from 40 to 200 ppb.

Table. Estimated sources and sinks of CO typical of the last decade (IPCC 1995).

Sources	Range (Tg(CO) yr⁻¹)
Technological	300-550
Biomass burning	300-700
Biogenics	60-160
Oceans	20-200
Methane oxidation (by OH)	400-1000
NMHC oxidation	200-600
Total sources	1800-2700
Sinks	
OH reaction	1400-2600
Soil uptake	250-640
Stratospheric loss	~100
Total sinks	2100-3000

- Carbonates:

CaCO₃ (calcium carbonate) is an important carbon-containing compound in rocks and in ocean sediments.

- Methane is the most abundant hydrocarbon in the atmosphere.

Table. Estimated sources and sinks of methane (IPCC 1995).

Identified sources	Individual estimate (Tg(CH ₄) yr ⁻¹)
Natural	
Oceans	10 (5-50)
Wetlands	115 (55-150)
Termites	20 (10-50)
Other	15 (10-40)
Total natural sources	160 (110-210)
Anthropogenic	
Fossil-fuel related sources:	
natural gas	40 (25-50)
coal mines	30 (15-45)
petroleum industry	15 (5-30)
coal combustion	? (1-30)
Total fossil-fuel related sources	100 (70-120)
Biospheric carbon:	
enteric fermentation	85 (65-100)
rice paddies	60 (20-100)
biomass burning	40 (20-80)
landfills	40 (20-70)
animal waste	25 (20-30)
domestic sewage	25 (15-80)
Total biospheric	275 (200-350)
Total anthropogenic sources	375 (300-450)
Total identified sources	535 (410-660)
Sinks	
Tropospheric OH	445 (360-530)
Stratosphere	40 (32-48)
Soils	30 (15-45)
Total sinks	515 (430-600)
Total global burden	4850 Tg (CH₄)

- Volatile organic compounds (VOCs) denote the entire set of vapor-phase atmospheric organics.

Table. Global biogenic VOC emission rate estimates by source (Tg/yr).

Source	Isoprene	Monoterpenes	Total VOC
Woods	372	95	821
Crops	24	6	120
Shrub	103	25	194
Ocean	0	0	5
Other	4	1	9
Total	503	127	1150

Table. Estimated global **anthropogenic** emission of nonmethane organic compounds.

Activity	Emission (Tg yr ⁻¹)
Fuel production/distribution:	
petroleum	8
natural gas	2
oil refining	5
gasoline distribution	2.5
Fuel consumption	
coal	3.5
wood	25
crop residues	14.5
charcoal	2.5
dung cakes	3
road transport	36
chemical industry	2
solvent use	20
uncontrolled waste burning	8
Other	
Total	142