

## **Lectures 33-34. Greenhouse gases and global warming.**

Objectives:

1. Greenhouse gases.
2. Global warming.

Readings: Turco: p. 365-386; Brimblecombe: p.184-189

### **1. Greenhouse gases.**

**Greenhouse effect** is a trapping of the earth's thermal radiation by the clouds and greenhouse gases (such as H<sub>2</sub>O, CO<sub>2</sub>, O<sub>3</sub>, CH<sub>4</sub>, etc.) and re-emitting it back to the surface.

- Various human activities result in additional loading of greenhouse gases with potential of global warming (or climate change).
- That the concentrations of greenhouse gases have been increasing over the past century is indisputable, a problem is how to identify their effects on climate.

Some history of global warming:

1896: Arrhenius proposed that continued coal combustion might increase the amount of CO<sub>2</sub> causing the global warming.

1938: G.Callendar stated that atmospheric CO<sub>2</sub> level increased by 10% since 1890s.

1955: G.Plass showed that CO<sub>2</sub> absorption lines don't overlap H<sub>2</sub>O absorption lines.

Thus adding of CO<sub>2</sub> to the atmosphere lead to interception of more infrared radiation.

1956: R.Revelle showed that, because of the chemistry of seawater, the ocean surface layer has limited uptake of CO<sub>2</sub>

1957: C. Keeling detected a rise of CO<sub>2</sub> after two years of measurements.

Table 33.1 Greenhouse gases whose concentrations have been increasing by human activities.

<b>Compound</b>	<b>Preindustrial concentrations (ppbv)</b>	<b>Present concentrations (ppbv)</b>	<b>Rate increase (%/yr)</b>	<b>Lifetime (yr)</b>
<b>CO<sub>2</sub></b>	<b>275,000</b>	<b>356,000</b>	<b>0.4</b>	<b>100-200</b> (hard to estimate)
<b>CH<sub>4</sub></b>	<b>800</b>	<b>1714</b>	<b>0.8</b>	<b>11</b>
<b>N<sub>2</sub>O</b>	<b>290</b>	<b>310</b>	<b>0.2</b>	<b>150</b>
<b>O<sub>3</sub></b> (in the troposphere)	<b>10</b>	<b>10-200</b>	<b>1</b>	<b>few hours</b>
<b>HALOCARBONS</b>				
<b>CFCl<sub>3</sub></b> (CFC-11)	<b>0</b>	<b>0.3</b>	<b>5</b>	<b>50</b>
<b>CF<sub>2</sub>Cl<sub>2</sub></b> (CFC-12)	<b>0</b>	<b>0.5</b>	<b>5</b>	<b>102</b>
<b>CF<sub>3</sub>Cl</b> (CFC-13)	<b>0</b>	<b>0.005</b>	<b>5</b>	<b>139</b>
<b>CF<sub>2</sub>HCl</b> (HCFC-22)	<b>0</b>	<b>0.1</b>	<b>7</b>	<b>13</b>
<b>CH<sub>3</sub>CCl<sub>3</sub></b> (methyl chloroform)	<b>0</b>	<b>0.16</b>	<b>7</b>	<b>5.4</b>
<b>CF<sub>3</sub>Br</b> (Halon-1301)	<b>0</b>	<b>0.002</b>	<b>20</b>	<b>65</b>
<b>C<sub>2</sub>F<sub>3</sub>Cl<sub>3</sub></b> (CFC-113)	<b>0</b>	<b>0.082</b>	<b>10</b>	<b>85</b>
<b>CF<sub>2</sub>Cl<sub>3</sub>CF<sub>2</sub>Cl</b> (CFC-114)	<b>0</b>	<b>0.02</b>	<b>5</b>	<b>300</b>
<b>C<sub>2</sub>F<sub>5</sub>Cl</b> (CFC-115)	<b>0</b>	<b>&lt; 0.01</b>	<b>5</b>	<b>1700</b>
<b>CH<sub>3</sub>Br</b> (methyl bromide)	<b>about 0</b>	<b>0.0015</b>	<b>15</b>	<b>1.5</b>

**CO<sub>2</sub> is the most important greenhouse gas.**

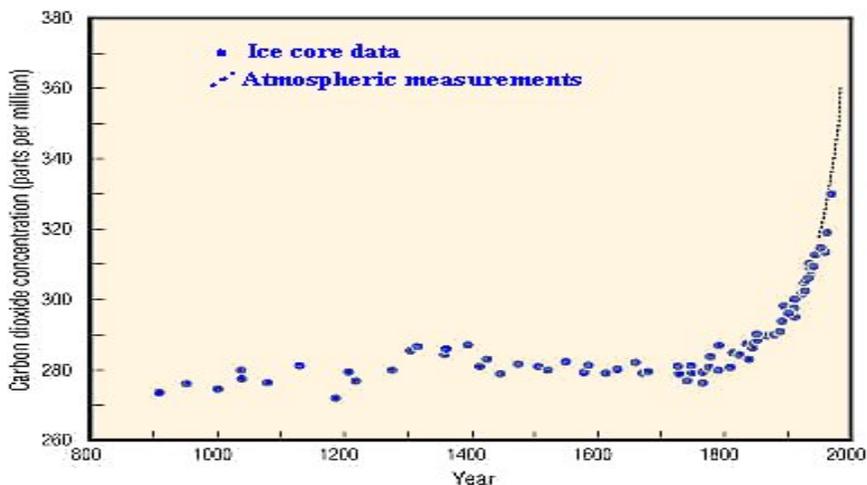
- Atmospheric lifetime for CO<sub>2</sub> is hard to determine because the cycling of carbon in the terrestrial biosphere and ocean occurs over timescales of decades to thousand of years.

Table 33.1 Global CO<sub>2</sub> budget (IPCC 1995).

	CO <sub>2</sub> (Gt(C) /yr )
<b>Sources</b>	
Fossil fuels combustion	5.5 +/- 0.5
Deforestation and land use changes	1.6 +/- 1.0
<b>Total known sources</b>	<b>7.1 +/- 1.1</b>
<b>Sinks</b>	
Oceanic uptake	2.0 +/- 0.8
Uptake by Northern Hemisphere forest regrowth	0.5 +/- 0.5
Retained in the atmosphere	3.2 +/- 0.2
<b>Total known sinks</b>	<b>5.7 +/- 1.0</b>
<b>Net imbalance</b>	<b>1.4 +/- 1.5</b>

- Accounting for the missing sink is a major research problem in the global CO<sub>2</sub> budget.

Figure 33.1 Long-term trend of atmospheric CO<sub>2</sub> concentrations.



**CH<sub>4</sub>**: shows a rapid growth since about 1900, and is currently increasing of about 1%/yr.

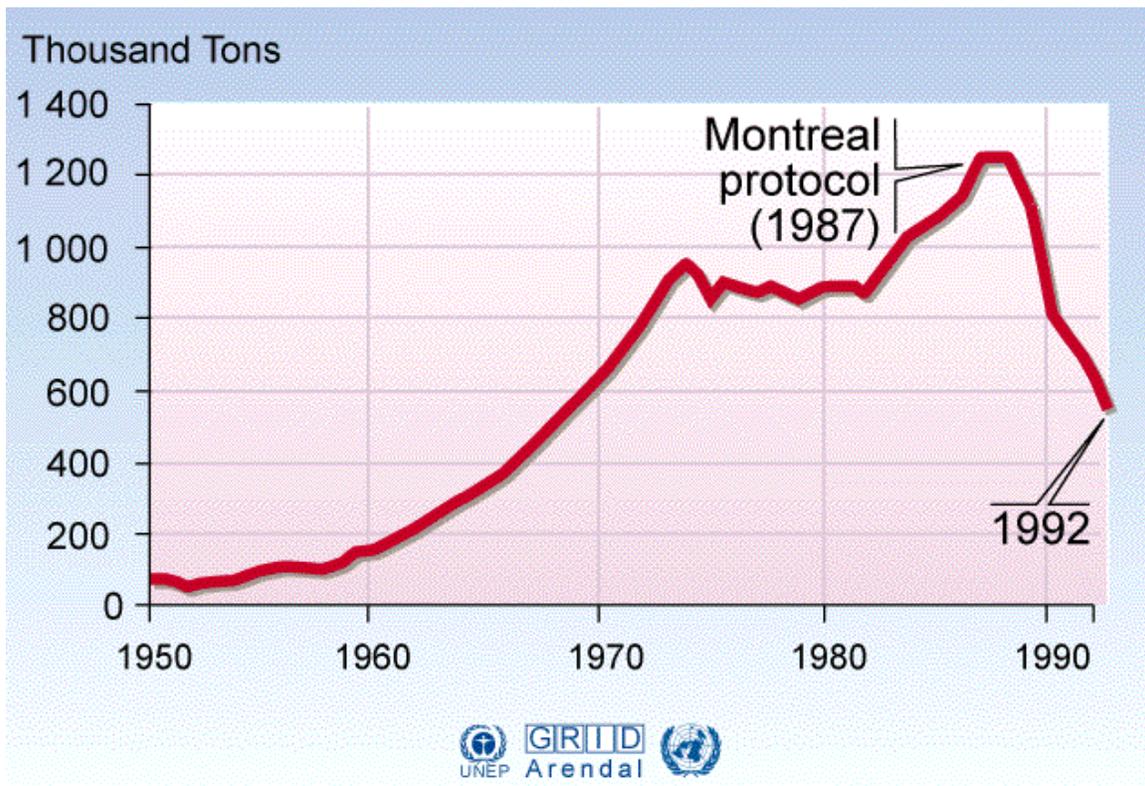
NOTE: Table (Lecture 12) shows natural and anthropogenic sources of CH<sub>4</sub>.

**N<sub>2</sub>O**: has currently lower concentrations, but shows steady increase of 0.2%/yr and has very long lifetime of 150 years.

NOTE: Table 11.1 (Lecture 11) shows natural and anthropogenic sources of N<sub>2</sub>O.

### **CFCs**

Figure 33.2 CFCs global production.



## 2. Global warming.

The **radiative forcing** is defined as the change in net irradiance (in  $\text{W m}^{-2}$ ) due to a change of a given species (for example, due to change in greenhouse gas concentration).

Estimation of the climate forcings from about 1850 to 1990 (IPCC, 1995):

$\text{CO}_2$ :  $+1.56 \text{ W m}^{-2}$

$\text{CH}_4$ :  $+0.47 \text{ W m}^{-2}$

$\text{N}_2\text{O}$ :  $+0.14 \text{ W m}^{-2}$

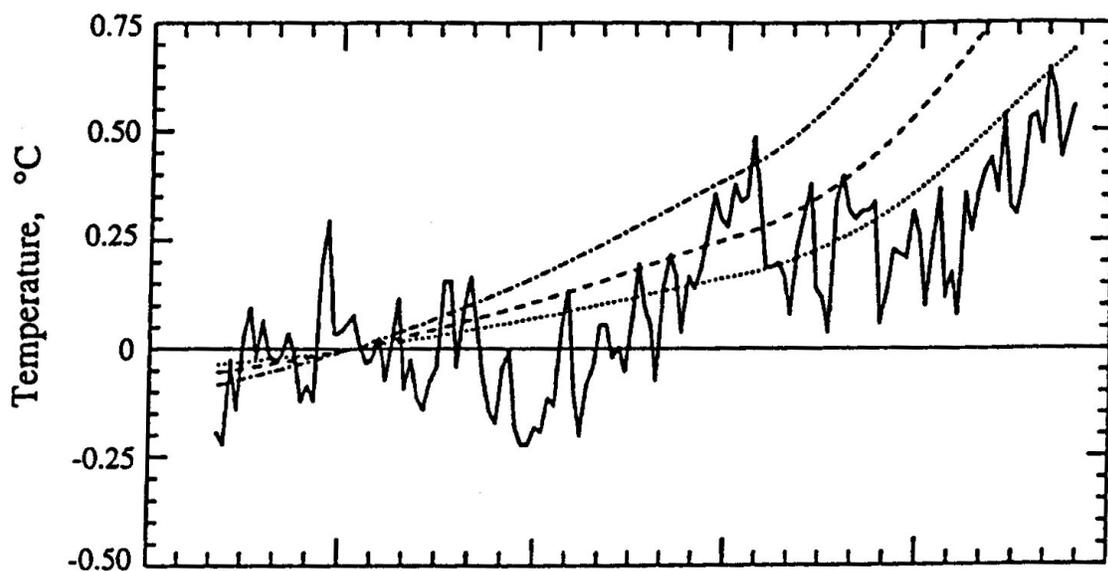
$\text{CFCl}_3$  (CFC-11):  $+0.06 \text{ W m}^{-2}$

$\text{CF}_2\text{Cl}_2$  (CFC-12):  $+0.14 \text{ W m}^{-2}$

( $\text{CCl}_4$  + HCFC-22 + CFC-113):  $+0.08 \text{ W m}^{-2}$

**TOTAL:  $+2.45 \text{ W m}^{-2}$**

Figure 33.3 Comparison of observed (solid line) and modeled global mean temperatures (dashed lines). Modeled temperatures were predicted by the model assuming  $\text{CO}_2$ -doubling scenario for three values of climate sensitivity.



Two major problems:

- 1) greenhouse gases give too much warming (about  $1.5^{\circ}$  to  $4.5^{\circ}\text{C}$  compare to the observed temperature change of about  $0.5^{\circ}$ );
- 2) radiative forcing due to greenhouse gases can not explain the pattern of global temperature change;

NOTE: accounting for the effects of anthropogenic aerosols we can explain better the observed temperature (will be discussed in Lectures 39-41).